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Numbers N, l, m, s

Used to describe an

electron in an atom n

Principle quantum

number Represents main

energy level of electron

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maximum # of electrons
in an Energy level = $2n^2$

Example: What is the
maximum number of
electrons that can be in
the 5th main energy
level? | The 2nd

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Describes the orbital
shape within an energy
level • number of

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energy level = n Orbital
Shapes designated s,p ...

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orientation of orbital in
space s ...

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number Represents main

energy level of _electron

maximum # of _electron

_ in an energy level _ =

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2n ² Example: What is the maximum number of electrons that can be in the 5th _ main energy level? | The 2 nd quantum number _____ Describes the orbital _____ shape within an ...

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acceleration due to

gravity = Example

problems: Rhoda Tiller

looks up to see a

flowerpot fall off a ledge

9.2 m above her head.

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