

Chapter 9 Review Stoichiometry Answers Section 3

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~~CHAPTER 9 REVIEW Stoichiometry~~
~~Stoichiometry b. Theoretically, how many moles of NH₃ will be produced? PROBLEMS Write the answer on the line to the left. Show all your work in the space provided. 1 88% The actual yield of a reaction is 22 g and the theoretical yield is 25 g. Calculate the percentage yield. 2. 6.0 mol of N₂ are mixed with 12.0 mol of H₂ according to the ...~~

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~~Stoichiometry. SECTION 2. PROBLEMS Write the answer on the line to the left. Show all your work in the space provided. 1. The following equation represents a laboratory preparation for oxygen gas: ... CHAPTER 9 REVIEW ...~~

~~CHAPTER 9 REVIEW~~
~~Chapter 9: Standard Review Worksheet 1. Answers will vary. An example is included below: 2H₂O₂(aq) → 2H₂O(l) + O₂(g) This describes the decomposition reaction of hydrogen peroxide. Microscopic: Two molecules of hydrogen peroxide (in aqueous solution) decompose to produce two molecules of liquid water and one molecule of oxygen gas.~~

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