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metal, the equilibrium
number of vacancies at 900 C
is $2.3 \times 10^{25} \text{ m}^{-3}$. If the
density and atomic weight of
this metal are 7.40 g/cm^3
and 85.5 g/mol ,
respectively, calculate the

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fraction of vacancies for
this metal at 900 C.

Solution This problem is
solved using two steps: (1)
calculate the total number
of lattice sites in silver,
N, using

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